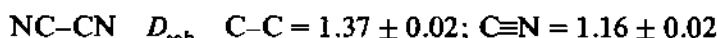


Errata

B. Corain, The Coordination Chemistry of Hydrogen Cyanide, Cyanogen and Cyanogen Halides, *Coord. Chem. Rev.*, 47 (1982) 165–200.

p. 170, Table 4, second line, should read:



Structure III does not contribute significantly and should be deleted. The sentence on p. 170 beginning: 'There is no doubt...', should read:

There is no doubt that a contribution from such a ground state would account for the particularly high dissociation energy of the C–C bond in C_2N_2 , as well as the short C–C bond distance, $1.37 \pm 0.02 \text{ \AA}$ [the sum of two $\text{C}(sp)$ covalent radii is $1.40 \text{ \AA}$].

A.B.P. Lever, Analytical Treatment of the Electronic Spectra of Some Low Symmetry Transition Metal Complexes, *Coord. Chem. Rev.*, 43 (1982) 63–85.

Equations (15)–(21) for the square pyramidal $\text{ML}_4\text{Z}^+\text{Z}^-$ species are only appropriate for the case where $e_\pi(\text{L}) = 0$.

In eqn. (A4), the ratio (1/990) should qualify the entire expression not simply the first term as presently indicated.

Eqn. (A9), is incorrect and should read:

$$\begin{aligned} DT &= -(7\sqrt{15} Dt)/2 \\ &= -(\sqrt{15}/20)[11\nu_3 - 3\nu_4 - (\nu_1 + \nu_5) - 2(\nu_2 + \nu_6) + 45B] \end{aligned}$$

Thus the expression for DT does depend upon the d^7 ground state.